

## Naming Binary Ionic Compounds Worksheet

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Write the name of each of the following compounds.

1.  $\text{CaBr}_2$  1. \_\_\_\_\_
2.  $\text{Al}_2\text{O}_3$  2. \_\_\_\_\_
3.  $\text{BaF}_2$  3. \_\_\_\_\_
4.  $\text{MgCl}_2$  4. \_\_\_\_\_
5.  $\text{AlCl}_3$  5. \_\_\_\_\_
6.  $\text{BaS}$  6. \_\_\_\_\_
7.  $\text{Na}_3\text{N}$  7. \_\_\_\_\_
8.  $\text{K}_3\text{P}$  8. \_\_\_\_\_

Write the chemical formula for the binary ionic compounds.

1. Magnesium nitride 1. \_\_\_\_\_
2. Calcium oxide 2. \_\_\_\_\_
3. Barium bromide 3. \_\_\_\_\_
4. Sodium nitride 4. \_\_\_\_\_
5. Potassium phosphide 5. \_\_\_\_\_
6. Cesium bromide 6. \_\_\_\_\_
7. Aluminum bromide 7. \_\_\_\_\_
8. Strontium bromide 8. \_\_\_\_\_
9. Aluminum oxide 9. \_\_\_\_\_