

# Polyatomic Ions Grade 10

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Name	Formula
Ammonium Ion	$\text{NH}_4^+$
Hydronium	$\text{H}_3\text{O}^+$
Hydroxide	$\text{OH}^-$
Carbonate	$\text{CO}_3^{2-}$
Bicarbonate (Hydrogen Carbonate)	$\text{HCO}_3^-$
Nitrate	$\text{NO}_3^-$
Nitrite	$\text{NO}_2^-$
Sulfate	$\text{SO}_4^{2-}$
Sulfite	$\text{SO}_3^{2-}$
Bisulfate (Hydrogen Sulfate)	$\text{HSO}_4^-$
Bisulfite (Hydrogen Sulfite)	$\text{HSO}_3^-$
Phosphate	$\text{PO}_4^{3-}$
Phosphite	$\text{PO}_3^{3-}$
Acetate	$\text{C}_2\text{H}_3\text{O}_2^-$
Perchlorate	$\text{ClO}_4^-$
Chlorate	$\text{ClO}_3^-$
Chlorite	$\text{ClO}_2^-$
Hypochlorite	$\text{ClO}^-$